



# Air Quality in stores, archives and exhibition spaces

Workshop on low energy climate control in museum and archives  
Copenhagen, 7 October 2010

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# The agenda

- On pollutants: compounds, sources, and their effects on materials
- How to measure
- Typical values observed in museum buildings
  
- Source control
- Control by ventilation
- Control by air cleaners
- Control by passive sorption

# Compounds and sources

- Ozone
- Nitrogen dioxide
- Sulphur dioxide
- Organic acids (acetic + formic)
- Volatile organic compounds
- Dust and particles



Damage caused by air pollution:  
*Lead corrosion*



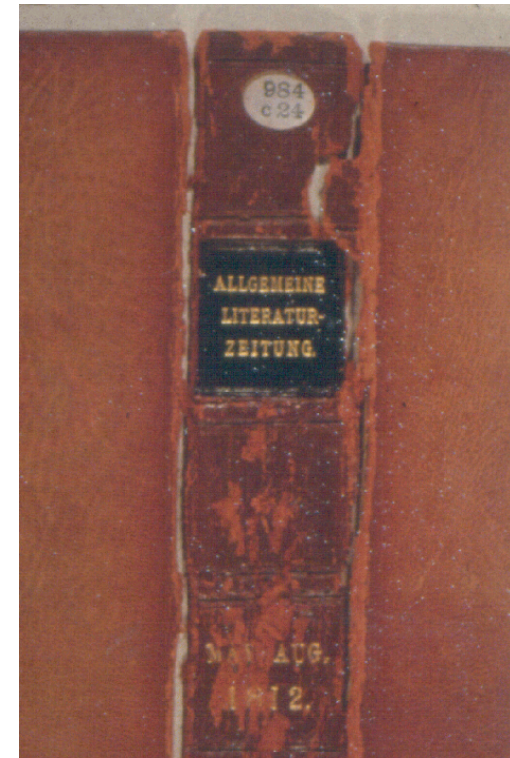
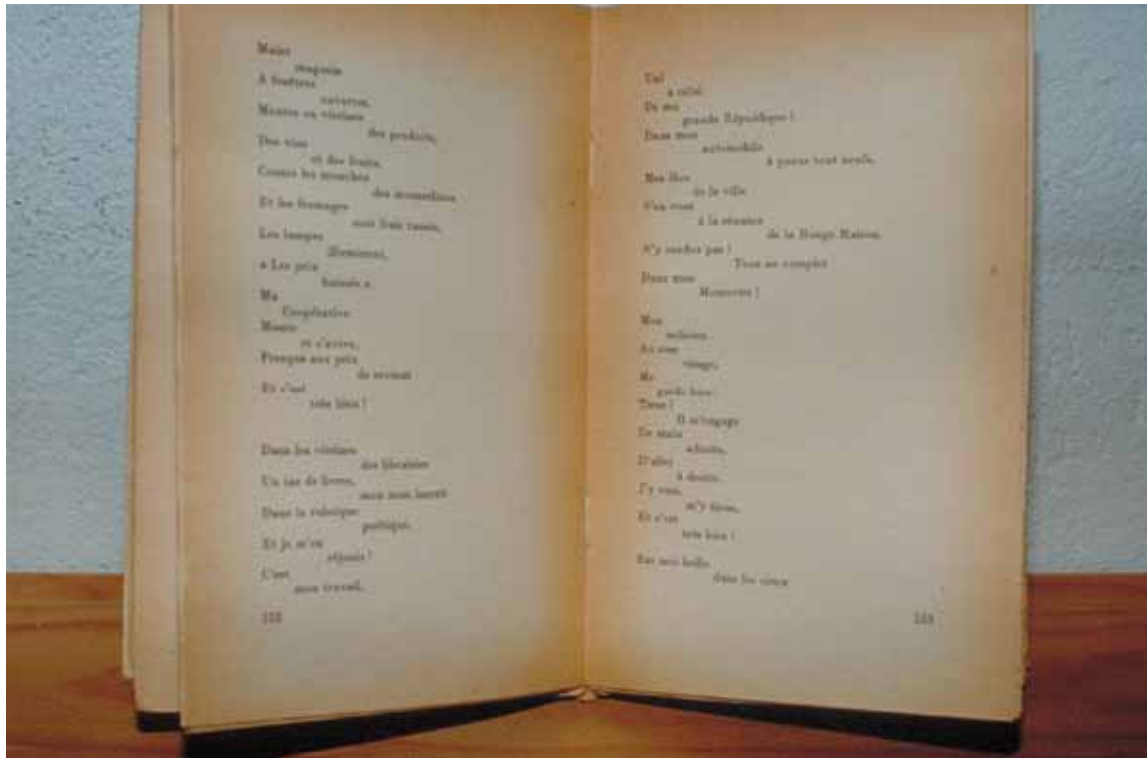
# Damage caused by air pollution: *Silver tarnish*



Damage caused by air pollution:  
*Oxidized rubber*



# Damage caused by air pollution: *Paper and leather*

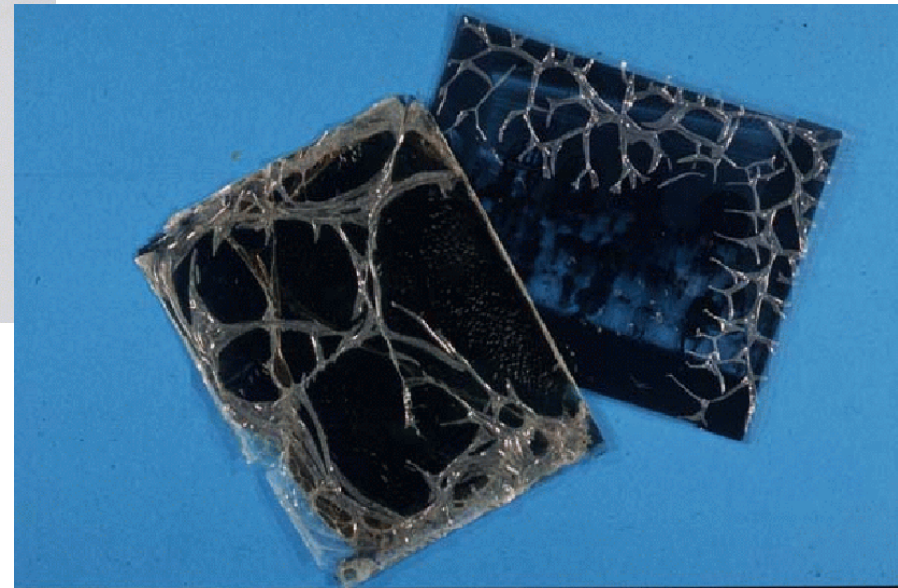


# Damage caused by air pollution: *Soiling*

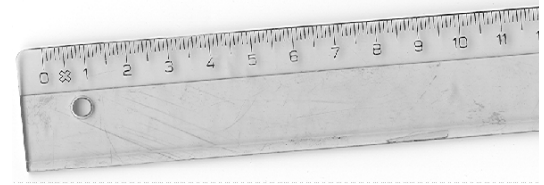




# Damage causing air pollution: *Deterioration of plastics*



# Monitoring

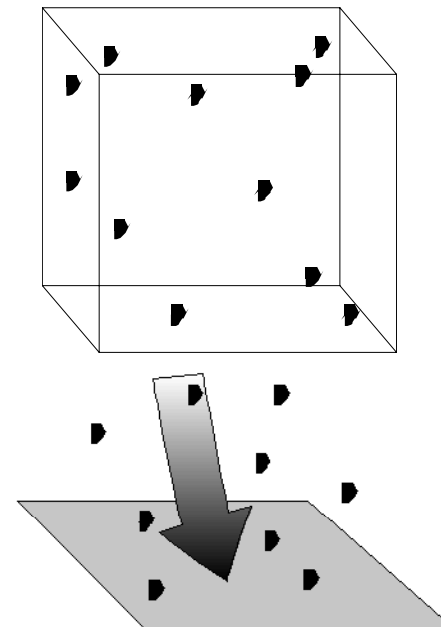


## Concentration measurements

- passive sampling, diffusion tubes
- active sampling, real-time instruments

## Dosimetry

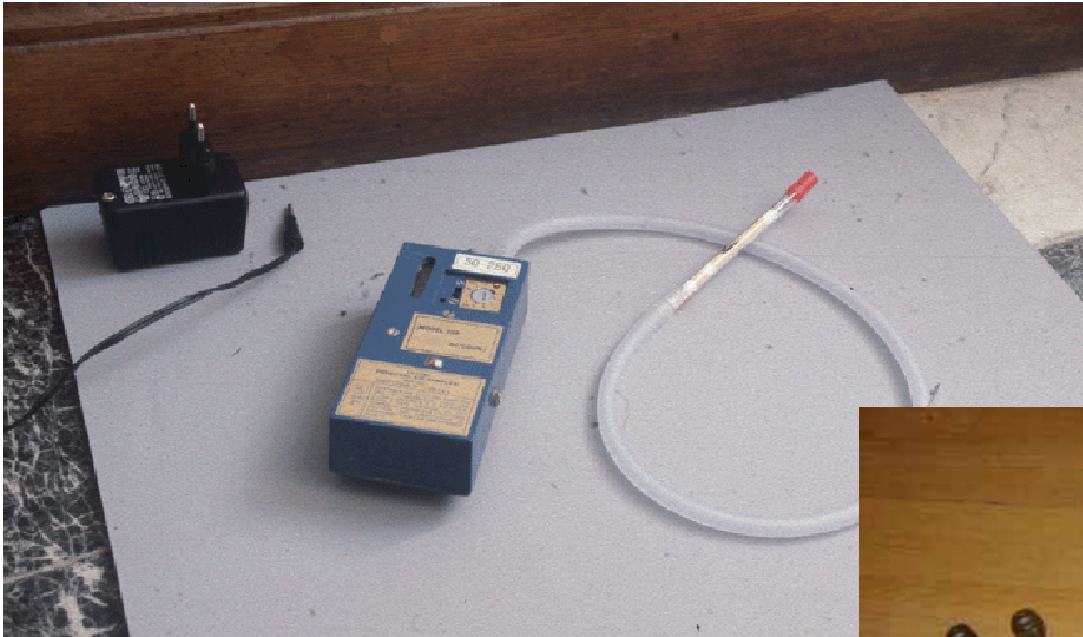
- "dummy" materials
- advanced dosimeters



# Diffusion tubes (passive)



# Active sampling



# Real-time monitoring

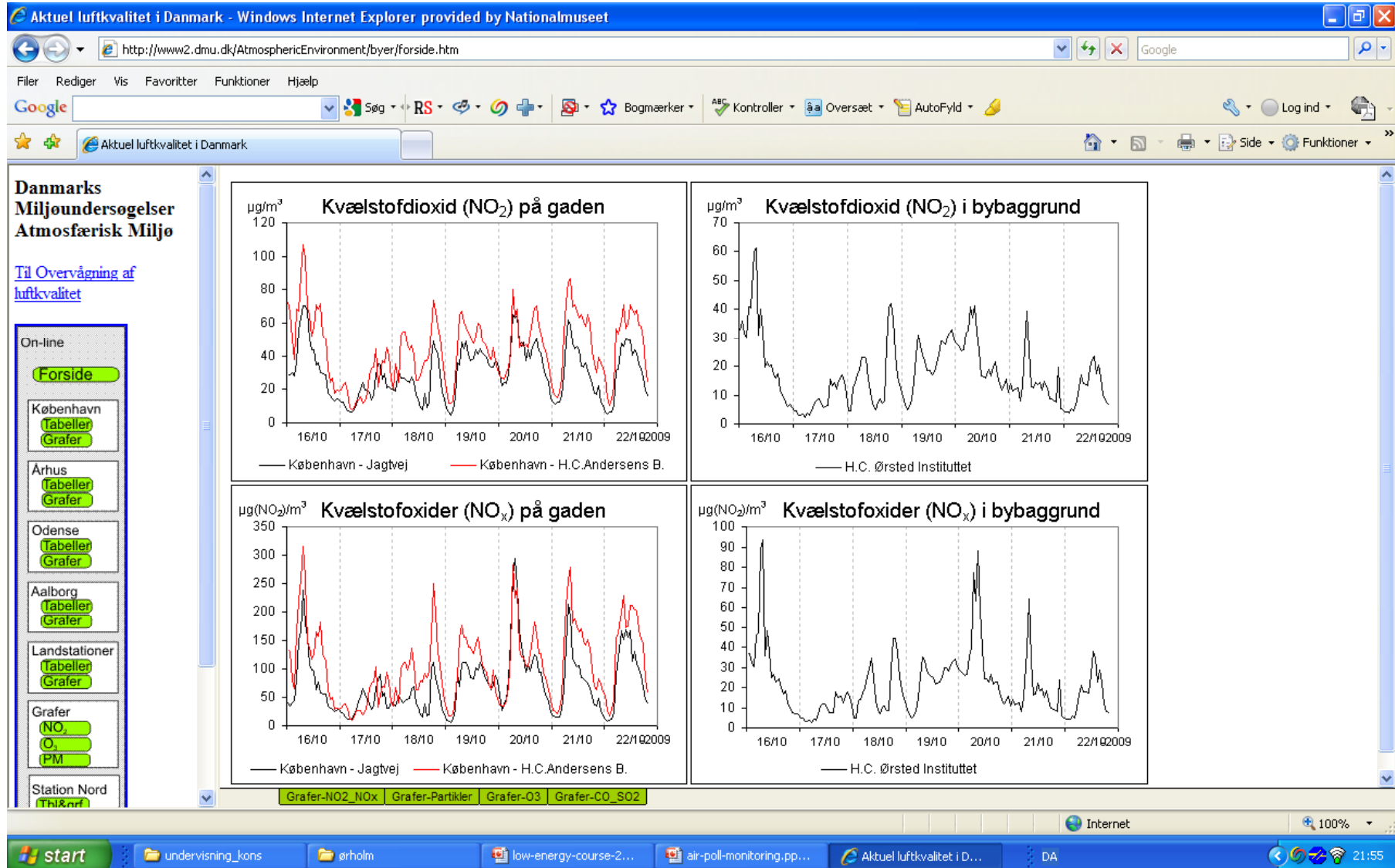


# Outdoor sampling

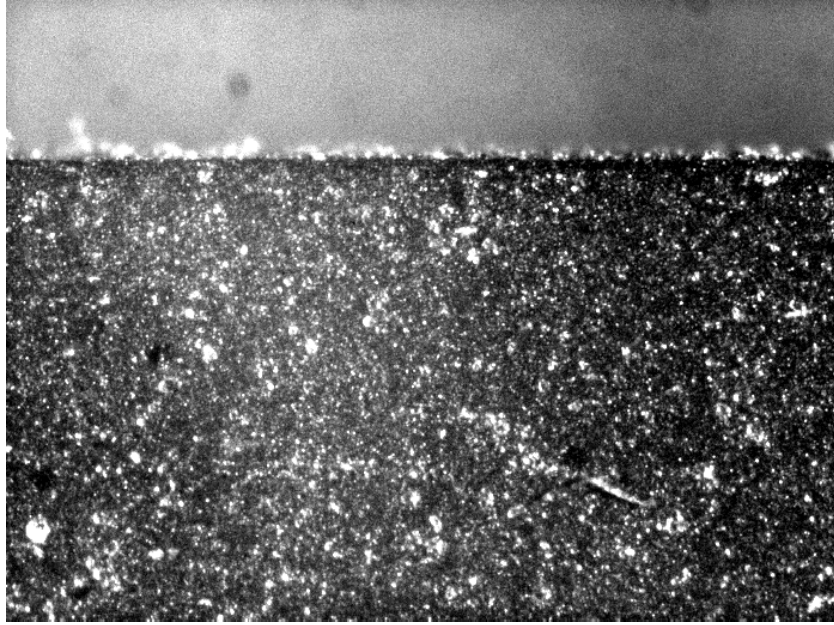




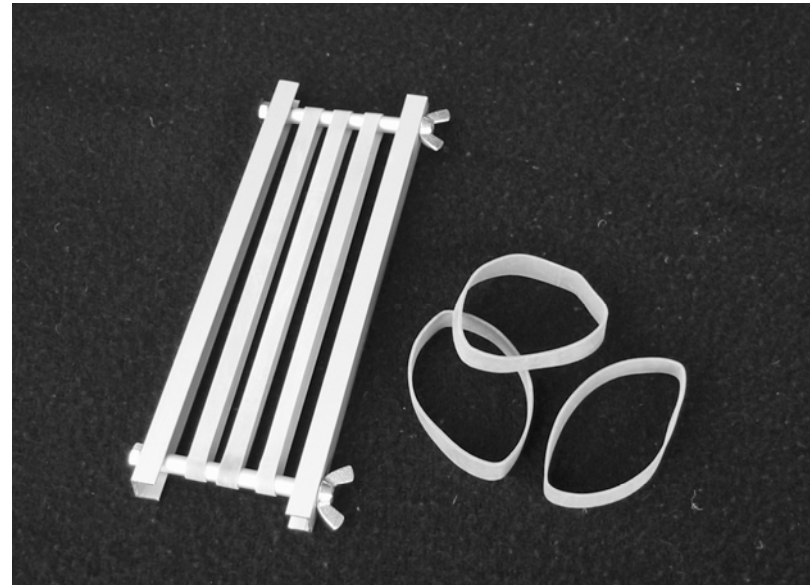
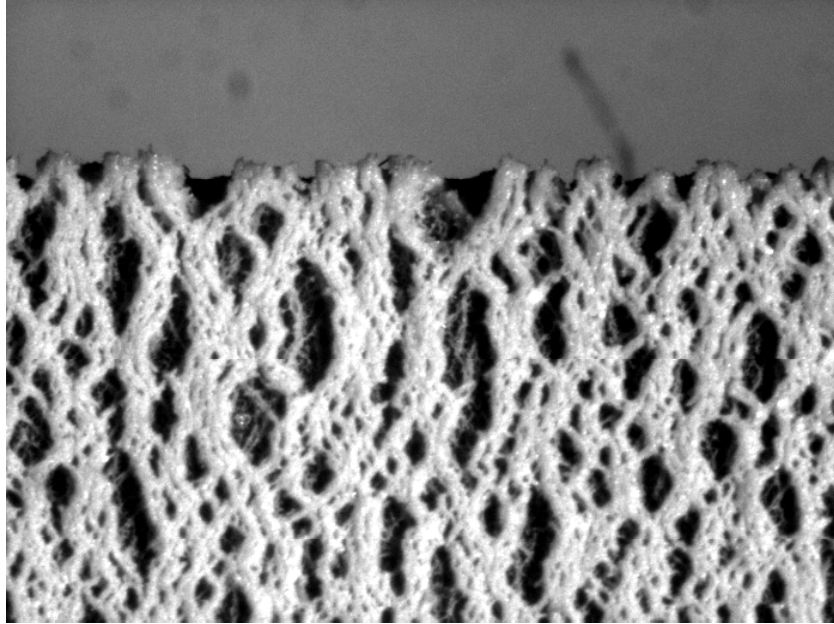
# Online data







Dosimetry:  
oxidizing agents  
(ozone)



# Dosimetry: corrosive environments



# Commercial dosimeters



# Commercial dosimeters

Purafil coupons  
(silver & copper)



NILU EWO (polymer)



# Examples of museum environments

**Odense Chatedral**  
**Urban environment**  
**Natural ventilation with heating**

Ozone outdoor: up to 45 ppb

**Ozone inside church: about 5 ppb**

**LOW**

**Ozone inside coffin: below detection**



Organic acids outdoor: 1-2 ppb

**Organic acids inside church: 1-5 ppb**

**LOW**

**Organic acids inside coffin: > 400 ppb**

**VERY HIGH**

# Examples of museum environments



**National Museum  
Ørholm Store "P"  
Suburban environment**

**Dehumidified, little heating**

**low air exchange rate  
(approx . 1 per day)**

Ozone outdoor: up to 50 ppb  
Ozone inside: **below 5 ppb**

**LOW**

Organic acids outdoor: 1 ppb  
Organic acids inside: **50-100 ppb**

**HIGH**

# Examples of museum environments



**National Museum  
Brede Store "921"  
Suburban environment**

**Full HVAC, high recirculation rate  
Carbon filtration**

Ozone outdoor: up to 50 ppb  
Ozone inside: **below 2 ppb**

**LOW**

Organic acids outdoor: 1 ppb  
Organic acids inside: **5-15 ppb**

**LOW**

# Examples of museum environments



Ozone outdoor: up to 70 ppb  
**Ozone inside: below 1 ppb**  
**LOW**

Organic acids outdoor: 1 ppb  
**Organic acids inside: 1-5 ppb**  
**LOW**

**Vejle Storage Facility**

**Internal re-circulation with dehumidification, no heating**  
**Rural environment**



# Control measures

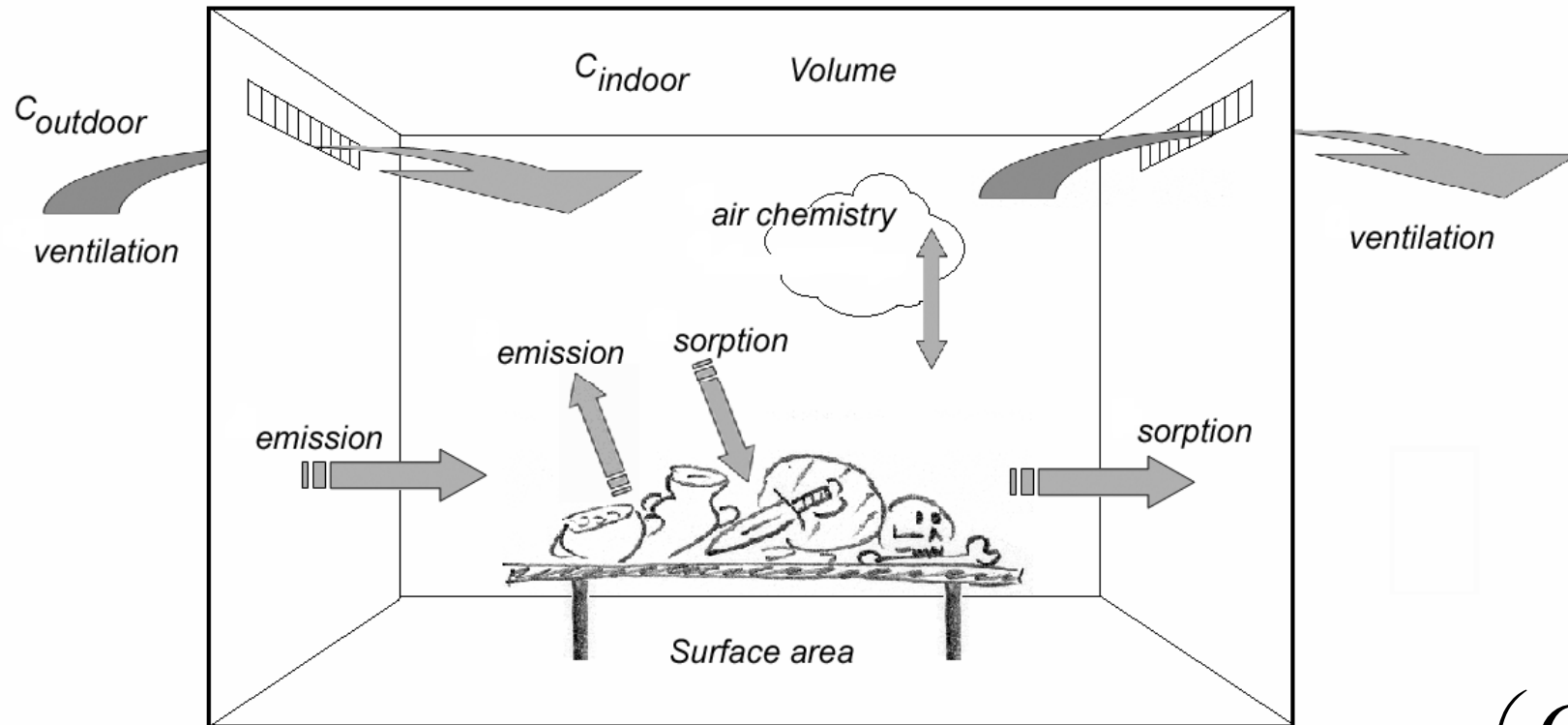
## **Outdoor pollutants: Block**

- Forced filtration
- Low air exchange
- Tortuous infiltration route

## **Indoor pollutants: Shift mass balance, remove from air**

- Avoid/remove/reduce source
- Increase reactive surface area
- Forced filtration (internal)
- Increase ventilation
  
- Decrease reaction rate (low temperature, low RH)

# Pollution pathways



$$C_i = \frac{C_o \times n}{n + S} + \frac{\left( \frac{G}{V} \right)}{n + S}$$

Ryhl-Svendsen, M., 'Indoor air pollution in museums: a review of prediction models and control strategies', *Reviews in Conservation* 7 (2006) 27-41.

# Block outdoor pollutants

Vejle Storage Building:  
No windows, few doors  
approx. one air exchange per day



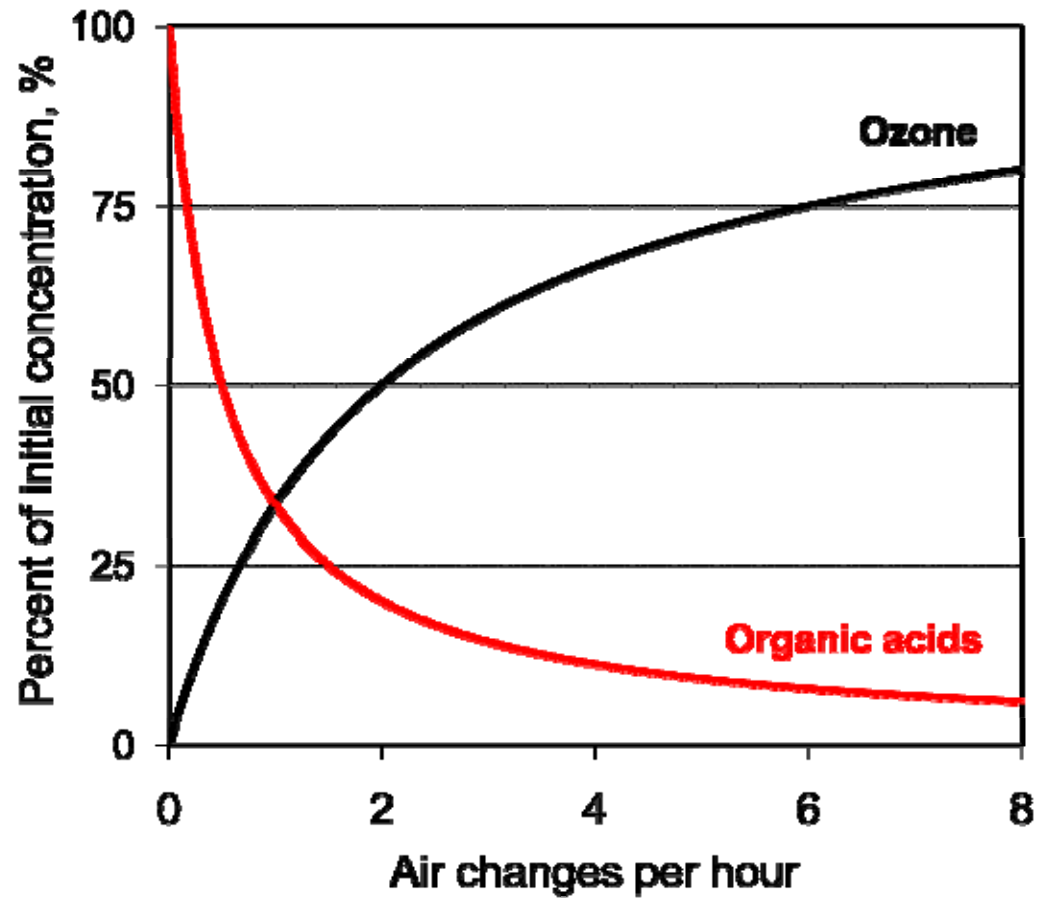
# Blocking pollutant's pathway: a double-edged sword



# Ventilation



# Dilution / infiltration



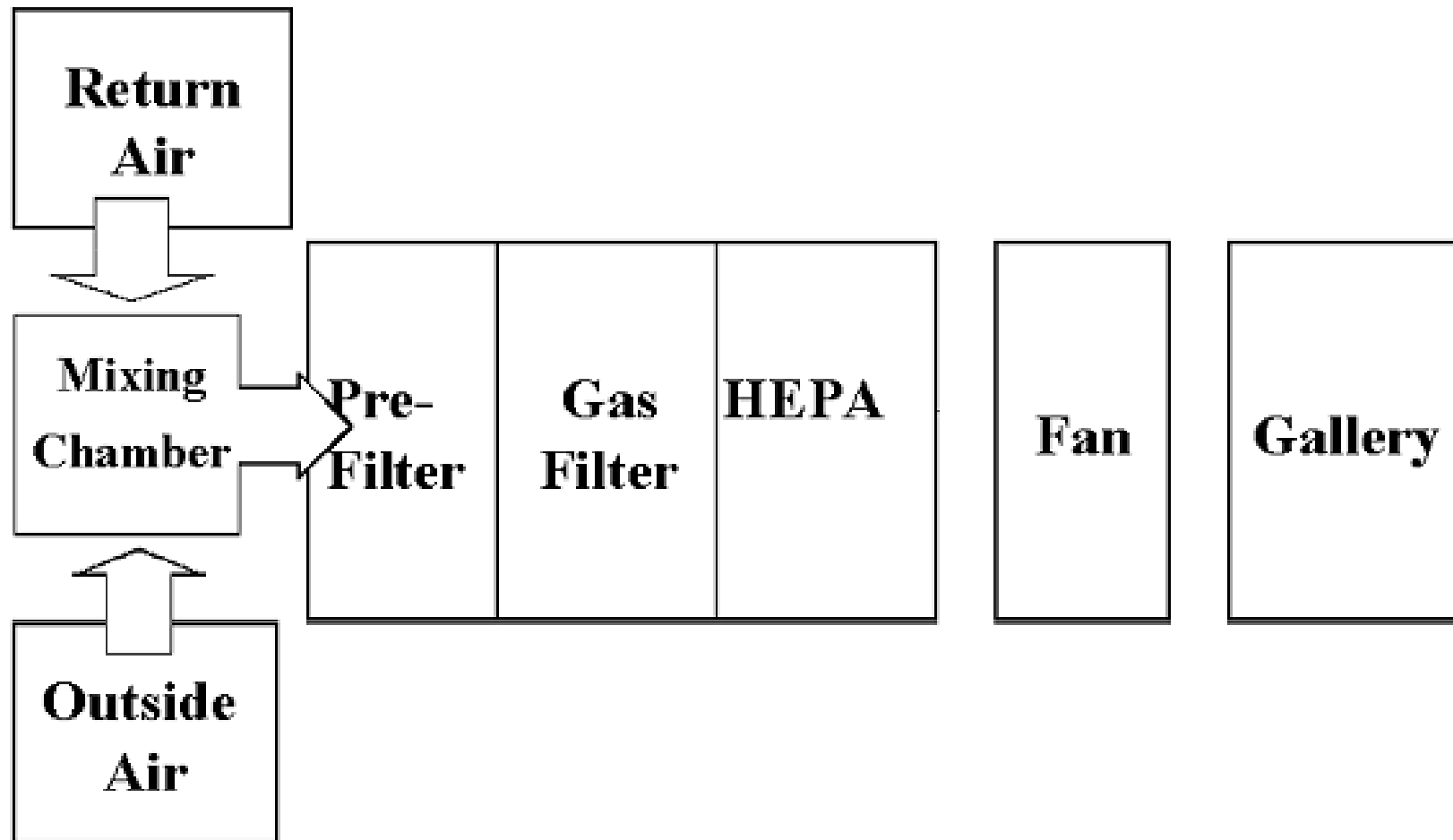
$$C_i = \frac{C_o \times n}{n + S} + \frac{\left(\frac{G}{V}\right)}{n + S}$$

# Reasons for ventilation or air handling?

- Heating and/or cooling
- Humidity control
- Providing "fresh" air
- Filtration?



# Air filtration





# Dust filters (bag type)





Activated carbon  
filters  
(charcoal granulates)

# Control of indoor generated pollutants



# Source control



# Source control



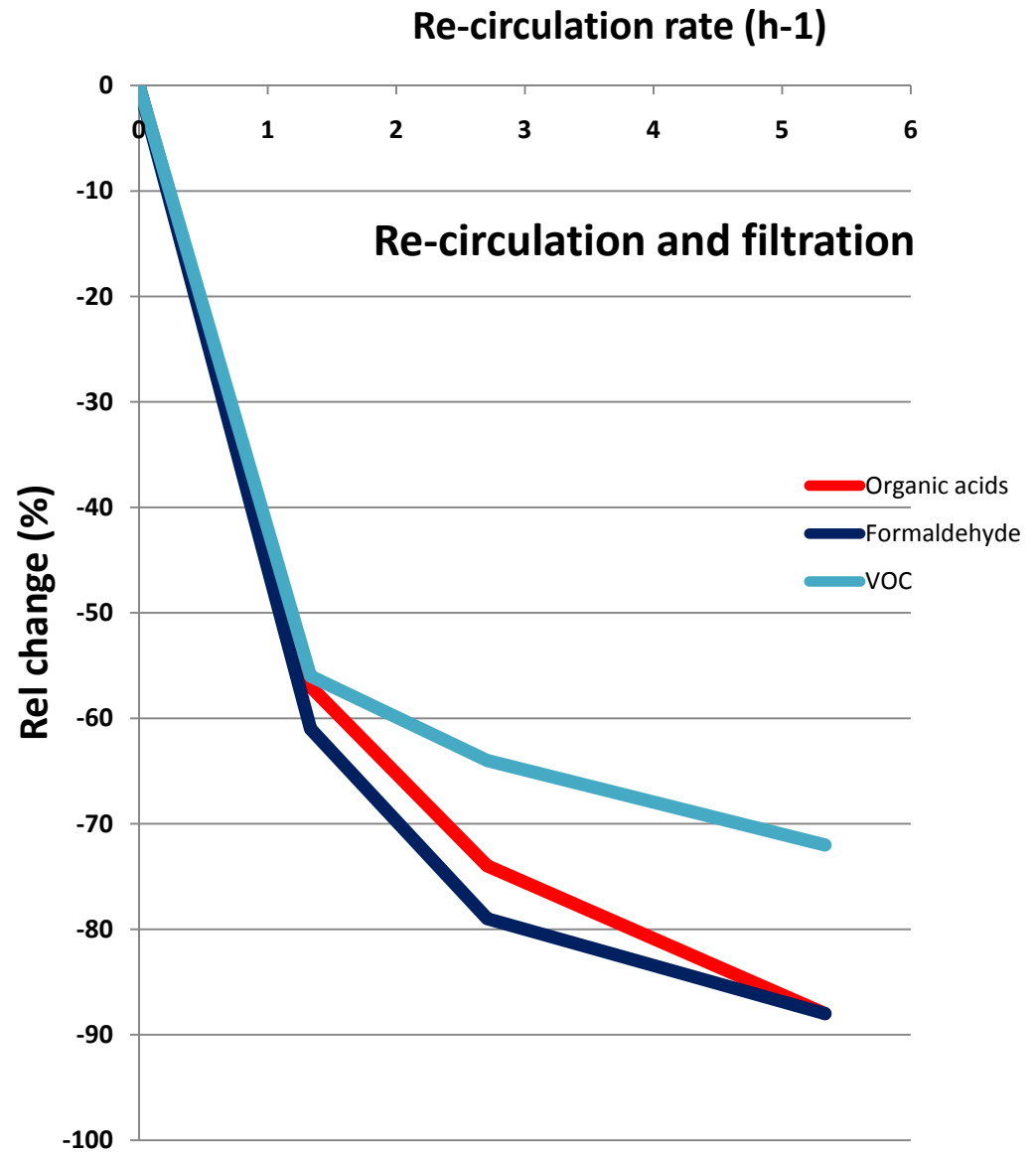
# Air cleaners



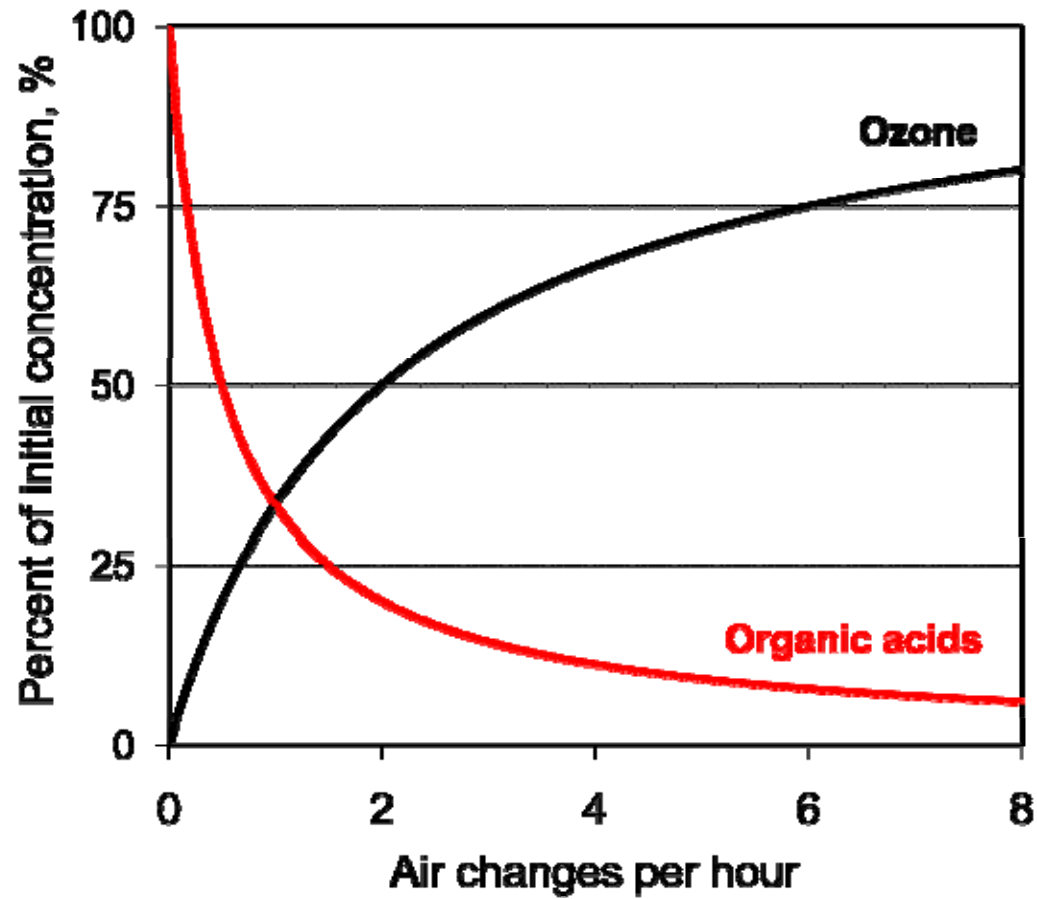


Energy consumption

80-1000 W depending on fan size and speed



# Dilution / infiltration





# Passive air cleaning (sorption)



# Air exchange rate and surface reactions: Outdoor pollutants

Surface removal rate

$$C_i = \frac{C_o \times n}{n + S}$$

Ozone:

- Office **1-4 h<sup>-1</sup>** (*Weschler, 2000*)
- Museum storage **2-3 h<sup>-1</sup>** (*Ryhl-Svendsen & Clausen, 2009*)
- Bedroom **>7 h<sup>-1</sup>** (*Weschler, 2000*)

Nitrogen dioxide:

- Museum gallery **0.4 h<sup>-1</sup>**
- Storage room **4.5 h<sup>-1</sup>** (*Blades et al, 2000*)

... and indoor generated pollutants

Surface removal rate

$$C_i = \frac{\left( \frac{G}{V} \right)}{n + S}$$

# Unfired brick (clay)



# Sorption on clay brick



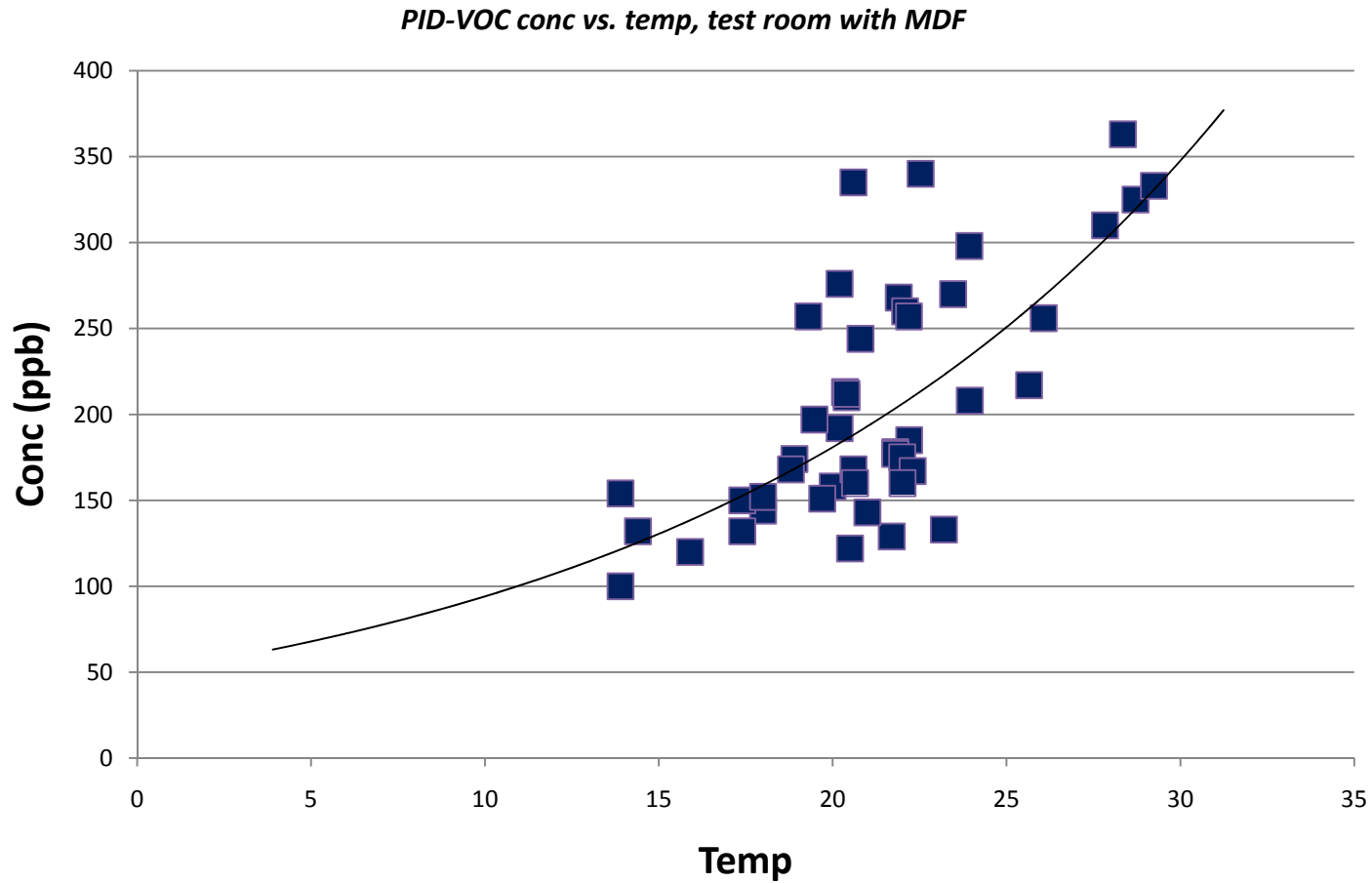
- 48 m<sup>3</sup> test room
- Approx. 20°C and 50% RH
- 0.5 m<sup>2</sup> clay wall per 1 m<sup>3</sup> room volume
- Low air exchange rate: 0.3 per hour
  
- **Organic acids: conc. decrease 30%**
- **Formaldehyde: conc. decrease 10%**
  
- Organic acid uptake (surface removal rate):  
**1.6 room volume per hour** (5x actual ventilation rate)

# Reducing reaction rate

Cold storage for unstable film materials



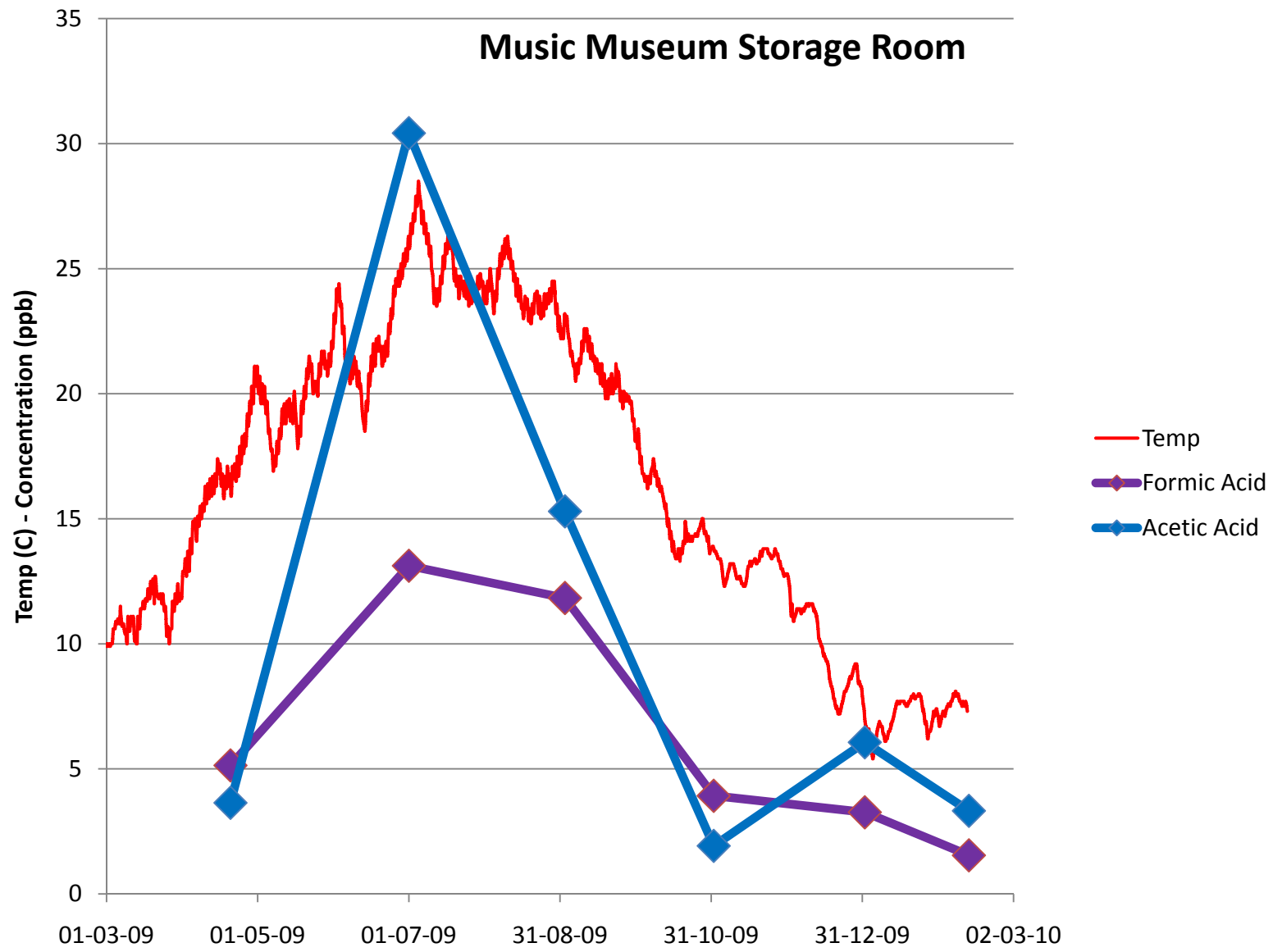
# Temperature and emission rate: VOCs from wood-fibre board







RH approx. 50%



# Micro-climates: boxes, books, drawers, paper stacks...

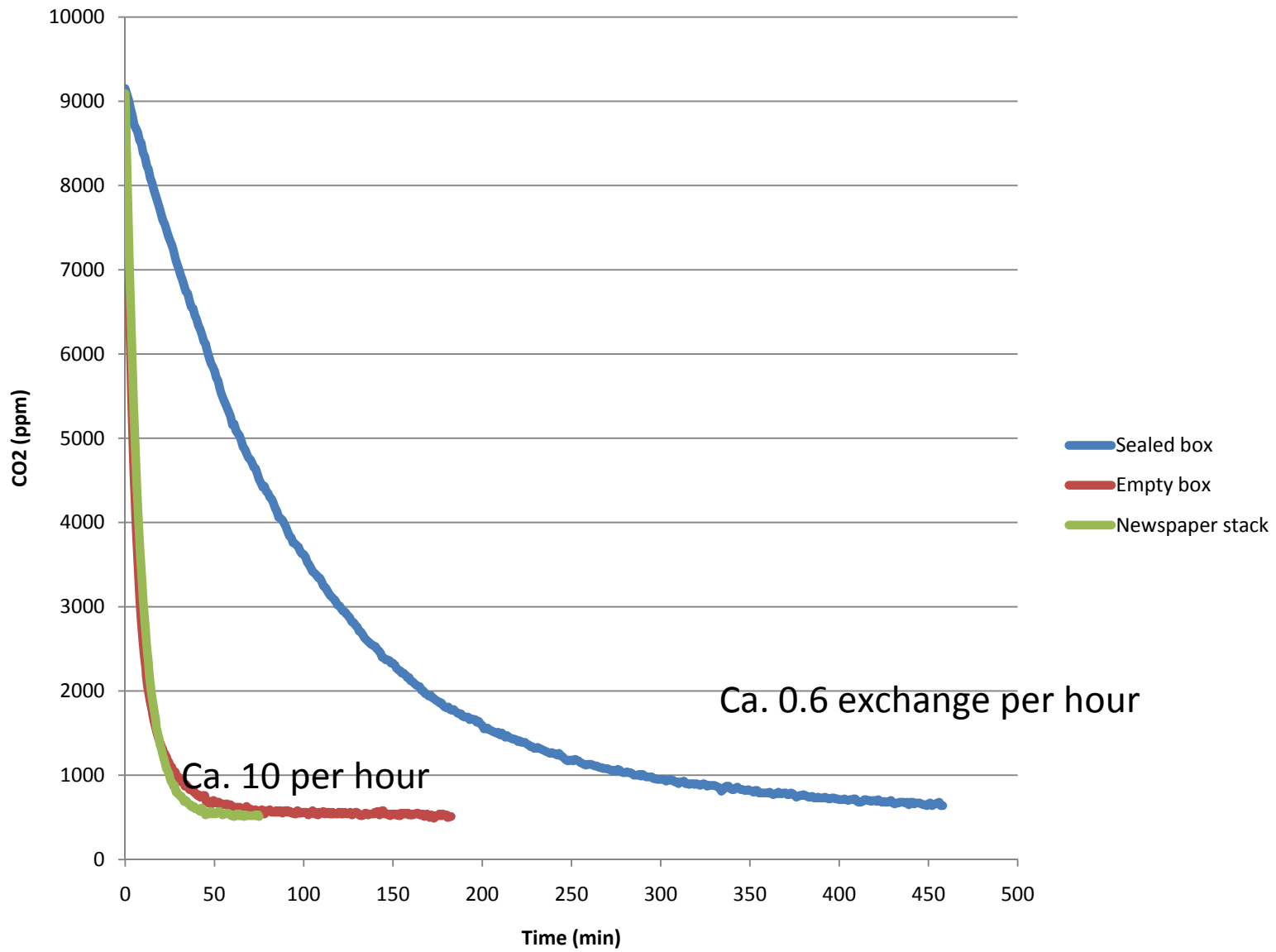


The air exchange of a box?  
Or a stack of paper?

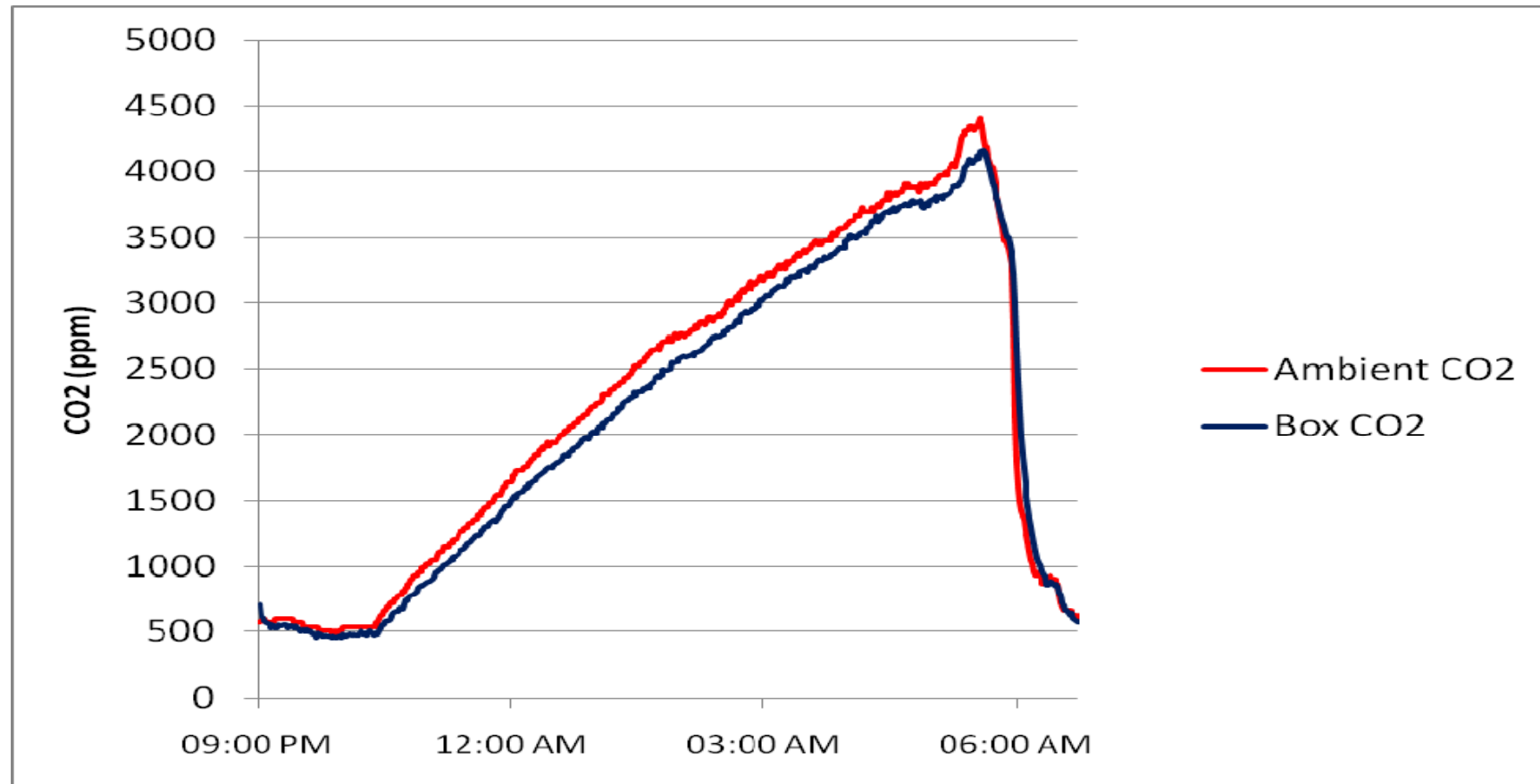


And what if the card board is sealed?





# Air diffusion through box board

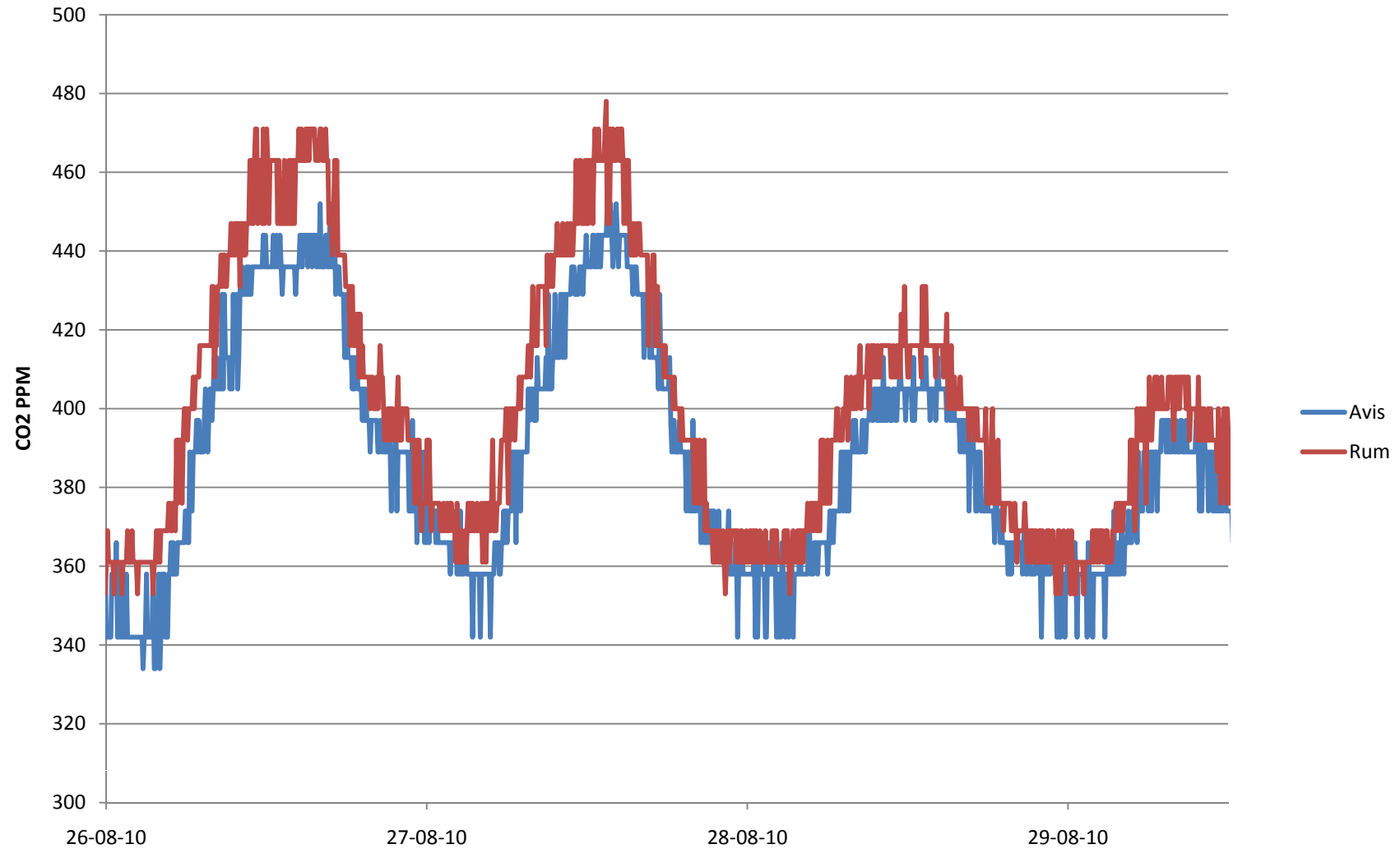




# Newspaper stack



# Air diffusion through newspaper stack





# Pollutants: diffusion and reaction



<b>Gas</b>	<b>Ambient conc. (room)</b>	<b>Level inside box (% of ambient)</b>
CO <sub>2</sub>	380 ppm	100%
NO <sub>2</sub>	6.9 ppb	60%
Organic acids	20 ppb	45%
O <sub>3</sub>	4.9 ppb	<10%